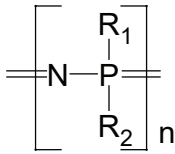
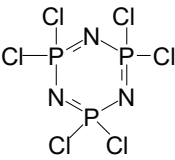


# PZ polyphosphazene

PARAMETER	UNIT	VALUE	REFERENCES
<b>GENERAL</b>			
Common name	-	polyphosphazene	
CAS name	-	poly[nitrilo[bis(2,2,2-trifluoroethoxy)phosphoranylidyne]]; poly[nitrilo(diphenoxyphosphoranylidyne)]	
Acronym	-	PZ	
CAS number	-	28212-50-2; 28212-48-8	
Formula			Allcock, H R, Current Opinion Solid State Mater. Sci., 10, 231-240, 2006.
<b>HISTORY</b>			
Person to discover	-	Allcock, H R	
Date	-	1965	
Details	-	Allcock's group published numerous papers and books on the subject	
<b>SYNTHESIS</b>			
Monomer(s) structure	-		
Monomer(s) CAS number(s)	-	940-71-6	
Monomer(s) molecular weight(s)	dalton, g/mol, amu	347.66	
Method of synthesis	-	thermal ring-opening polymerization of hexachlorophosphazene followed by esterification of the intermediate poly(dichlorophosphazene) with either amines or sodium salts of alcohols	
Temperature of polymerization	°C	250	
Time of polymerization	h	20	
Number average molecular weight, $M_n$	dalton, g/mol, amu	132,000-173,000	
Mass average molecular weight, $M_w$	dalton, g/mol, amu	10,000-5,000,000	Nakamura, H; Masuko, T; Kojima, M; Magill, J H, Macromol. Chem. Phys., 200, 2519-24, 1999.
<b>STRUCTURE</b>			
Crystallinity	%	19.96-52.35	Hazendonk, P; deDenus, C; Iuga, A; Cahoon, P; Nilsson, B; Iuga, D, J. Inorg. Organometal. Polym. Mater., 16, 4, 343-57, 2006.
Cell type (lattice)	-	orthorhombic ( $\alpha$ and $\gamma$ ), monoclinic ( $\beta$ )	Hazendonk, P; deDenus, C; Iuga, A; Cahoon, P; Nilsson, B; Iuga, D, J. Inorg. Organometal. Polym. Mater., 16, 4, 343-57, 2006.
Cell dimensions	nm	a:b:c=1.014:0.935:0.486 ( $\alpha$ ); a:b:c=2.06:0.94:0.486 ( $\gamma$ ); a:b:c=1.003:0.937:0.486 ( $\beta$ ); a:b:c=1.19:1.19:0.486 (hexagonal mesophase)	Hazendonk, P; deDenus, C; Iuga, A; Cahoon, P; Nilsson, B; Iuga, D, J. Inorg. Organometal. Polym. Mater., 16, 4, 343-57, 2006.

# PZ polyphosphazene

PARAMETER	UNIT	VALUE	REFERENCES
<b>Polymorphs</b>	-	$\alpha, \beta, \gamma$	Hazendonk, P; deDenus, C; Iuga, A; Cahoon, P; Nilsson, B; Iuga, D, J. Inorg. Organometal. Polym. Mater., 16, 4, 343-57, 2006.
<b>PHYSICAL PROPERTIES</b>			
<b>Density at 20°C</b>	g cm <sup>-3</sup>	1.63-1.74	
<b>Refractive index, 20°C</b>	-	1.6-1.75	
<b>Melting temperature, DSC</b>	°C	110-249	
<b>Thermal expansion coefficient, 23-80°C</b>	10 <sup>-4</sup> °C <sup>-1</sup>	1.9-10.7	Nakamura, H; Masuko, T; Kojima, M; Magill, J H, Macromol. Chem. Phys., 200, 2519-24, 1999.
<b>Glass transition temperature</b>	°C	7 to -82	Hazendonk, P; deDenus, C; Iuga, A; Cahoon, P; Nilsson, B; Iuga, D, J. Inorg. Organometal. Polym. Mater., 16, 4, 343-57, 2006; Allegra, G; Meille, S V, Macromolecules, 37, 3487-96, 2004.
<b>Surface free energy</b>	mJ m <sup>-2</sup>	15.8	
<b>CHEMICAL RESISTANCE</b>			
<b>Good solvent</b>	-	water (some)	
<b>PROCESSING</b>			
<b>Applications</b>	-	flame retardation, fuel cells, fuel hoses, medical (bone regeneration, stomach wall regeneration), membranes, waterproofing	
<b>Outstanding properties</b>	-	thermal stability	
<b>BLENDS</b>			
<b>Suitable polymers</b>	-	PDMS, PI, PLA, PTFE (composite)	
<b>ANALYSIS</b>			
<b>NMR (chemical shifts)</b>	ppm	C, H, F, and P data included in ref.	Hazendonk, P; deDenus, C; Iuga, A; Cahoon, P; Nilsson, B; Iuga, D, J. Inorg. Organometal. Polym. Mater., 16, 4, 343-57, 2006.
<b>x-ray diffraction peaks</b>	degree	diffraction data in ref.	Tang, H; Pintauro, P N, Eur. Polym. J., 35, 1023-35, 1999.